

INTERCELLULAR DIFFUSION OF MACROPHAGE-DERIVED NITRIC OXIDE AND PEROXYNITRITE.

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Macrophage-derived nitric oxide ($\cdot\text{NO}$) has cytotoxic effects mediated either directly by $\cdot\text{NO}$ or its derived species such as peroxynitrite, arising from the reaction of $\cdot\text{NO}$ with superoxide radical ($\text{O}_2^{\cdot-}$). To propose a predominant $\cdot\text{NO}$ -dependent cytotoxicity mechanism, the different diffusion properties of $\cdot\text{NO}$ and peroxynitrite should be considered as they will affect their radii of action. Herein, we evaluated how the diffusion of $\cdot\text{NO}$ derived from activated RAW 264.7 macrophages is modulated by the simultaneous formation of $\text{O}_2^{\cdot-}$. Our experimental model consists in the co-incubation of differentially activated macrophages for the production of $\cdot\text{NO}$, $\text{O}_2^{\cdot-}$ or both (and hence peroxynitrite) with red blood cells (RBC) as targets. Diffusion of $\cdot\text{NO}$ /peroxynitrite to the RBC was evaluated as intracellular oxyhemoglobin oxidation and nitrosylhemoglobin formation. Our results showed that: i) oxyhemoglobin oxidation yields obtained with $\cdot\text{NO} + \text{O}_2^{\cdot-}$ are smaller than with only $\cdot\text{NO}$ and are dependent on diffusion distances; ii) nitrosylhemoglobin was partially decreased when macrophages also produced $\text{O}_2^{\cdot-}$, indicating that diffusion of $\cdot\text{NO}$ to RBC only partially outcompetes peroxynitrite formation and iii) macrophage-derived peroxynitrite can diffuse and reach target cells. Computer assisted-simulations were performed and are on line with the experimentally-obtained results. Our data supports an *in vivo* scenario where the cytotoxic effects of $\cdot\text{NO}$ will be strongly influenced by the intercellular diffusion distances and by the concomitant formation of $\text{O}_2^{\cdot-}$.